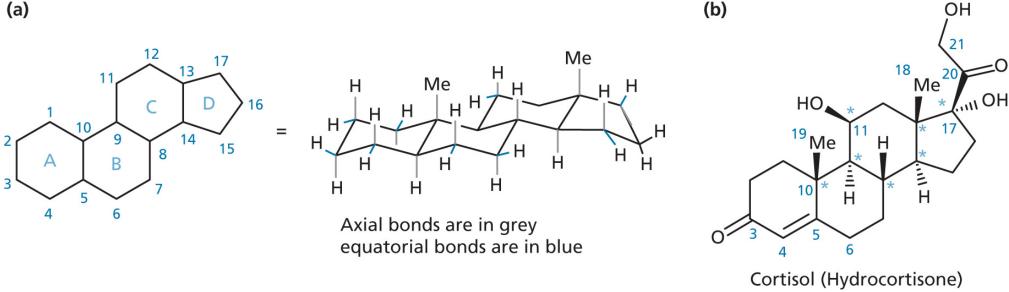


Topic 10: Steroidal antiinflammatory agents

Introduction to steroids

- Steroids are important endogenous hormones found in many life forms.
- They all share a common <u>tetracyclic structure</u>, as shown in the figure, but they <u>vary</u> in the substituents and functional groups that are present.
- The stereochemistry of the rings in fully saturated steroids is identical in mammalian steroids, where the three 6-membered rings have chair conformations.
- There are several asymmetric centers present, but only one stereoisomer occurs naturally for any particular steroid.

• Cortisol has seven asymmetric centers, but only the stereoisomer shown in the figure exist naturally



- One of the terminology used in the nomenclature of steroids is the substituents are often described as being alpha (α) or beta (β).
- α -substituents are below the general *plane* of the steroid and are presented by hatched wedges in two-dimensional diagrams (e.g. H9 & H14), whereas β -substituents are above the *plane* and are presented by solid wedges (e.g. C18 & C19).

- The position of double bonds in steroids is usually identified by the symbol delta (Δ); e.g. Δ^4 signifies the double bond between C4 & C5 in cortisol.
- If there is any doubts then numbers of both carbons are indicated, e.g. cholesterol has a double bond between C5 & C6, rather than C5 & C10, so this is indicated as $\Delta C^{5(6)}$

- Steroids are <u>hydrophobic compounds</u> owing to their extensive hydrocarbon skeleton. This is an important character as the hormonal steroids have to cross cell membranes in order to interact with intracellular steroid receptors.
- All of the important endogenous steroids have **polar functional groups**, such as alcohols, phenols, and ketones. These play a crucial role in the binding steroids to their target receptors, but their presence does not alter the hydrophobic nature of the molecule as a whole.
- Because most steroids are **hormones**, they are present in very small quantities in the body. The exception is cholesterol which is present in much larger quantities and has a non-hormonal roles " in developing cholesterol-lowering drugs".

- Here, we will concentrating on those steroid released from the adrenal cortex of the adrenal gland- the **adrenocorticoids**.
- There are two types of adrenocorticoids- the **glucocorticoids** and the **mineralocorticoids**.
- Glucocorticoids act on carbohydrate, fat and protein metabolism mainly in liver, muscles, and brain cells. They have an important anti-inflammatory effect which is separated from their metabolic effects.
- Mineralocorticoids regulate electrolyte balance through sodium ion retention in kidney cells.
- The major endogenous glucocorticoids are corticosterone, cortisone and cortisol "also known as hydrocortisone".

- Aldosterone is the major endogenous mineralocorticoid.
- The adrenocorticoids are examples of steroids having <u>a pregnane</u> skeleton-steroids having a two carbon side chain at position 17 of the tetracyclic steroid skeleton.

• An imbalance of these steroids can lead to certain diseases. E.g. an excess of glucocorticoids causes **Cushing's syndrome**, where as a deficit results in **Addison's disease**. An excess of mineralocorticoids leads to **Conn's syndrome**.

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- The glucocorticoids have an important clinical role in replacement therapy for Addison's disease.
- They have also been used as anti-inflammatories and immunosuppressants in the treatment of conditions such as asthma, hypersensitivity, rheumatoid arthritis, cancer and diseases which have an autoimmune or inflammatory effect.
- One of the most important applications of glucocorticoids in medicine is as anti-inflammatory agents.
- Unfortunately, the endogenous glucocorticoids suffer from the fact that they have **mineralocorticoid** and **immunosuppressant effects**, which can cause oedema and increase susceptibility to infection.
- Moreover, the endogenous glucocorticoids affect a large number of enzymes in different cell types.

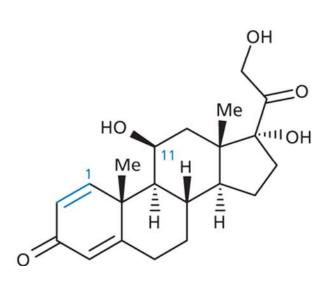
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Orally active analogues of cortisol

- In 1947, it was found that cortisone could relive the symptoms of rhuematoid arthritis.
- Cortisone is readily converted in the liver to cortisol and it is now thought that the effects of cortisone are actually due to cortisol.
- A large number of analogues have been synthesized which have identified the features of cortisol that are important for corticosteroid activity.
- In essence, all the functional groups are important, and the removal of any of these groups either reduces or eliminates activity.

- However, further studies have shown that the introduction of extra substituents can **increase** activity, which **allows the removal** of one of the original functional groups.
- Introducing a 9α -fluoro substituent to give <u>fludrocortisone</u> increased activity 10-fold, but it also increased minieralocorticoid activity 300-600 times.
- the introduction of an extra double bond at the Δ^1 position increased activity 4-fold without increasing mineralocoticoid activity (**prednisolone and prednisone**).
- Introducing substituents such as methyl or flourine at the 6α -**position** has also been found to be beneficial because these groups serve to block metabolism at that position. E.g. **methylprednisolone** has a 6α -methyl group.

Methylprednisolone



Prednisolone

- A methyl group was introduced at **C-16** to see whether it would block the metabolic reduction of the C-20 keto group of hydrocortisone analogues- a reaction that is known to lead to inactive metabolites.
- Further research revealed that the introduction of C-16 substituents such as a methyl or hydroxyl group, **counteracted** the mineralocorticoid effect of a 9-fluoro substituent.
- This resulted in the development of <u>triamcinolone</u>, <u>dexamethasone</u>, <u>betamethasone</u>, and <u>flumetasone</u> <u>pivalate</u>, all of which have increased glucocorticoid activity and negligible mineralocorticoid side effects.

Triamcinolone

Dexamethasone

Betamethasone

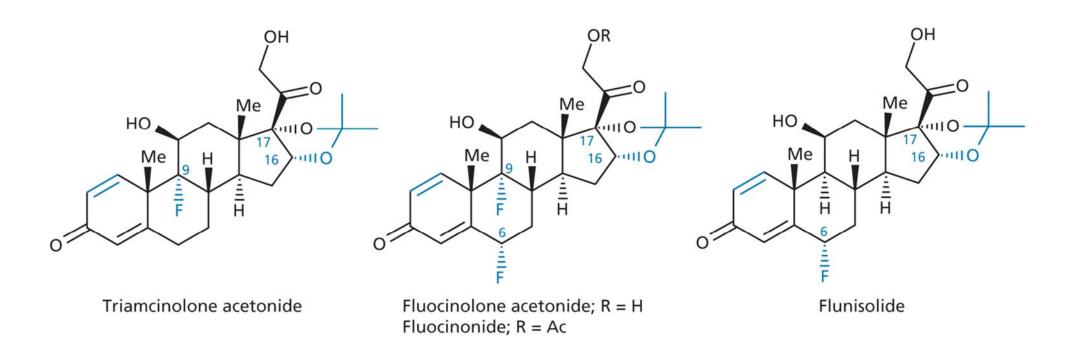
Flumetasone pivalate

* Topical glucocorticoids as antiinflammatory agents

☐ Cortisol analogues

- Glucocorticoids are often applied topically to treat skin inflammation.
- <u>Triamcinolone acetonide</u> is one such agent; the acetonide group links the alcohol substituents at C16 & C17 of triamcinoline thus reducing the polarity of the molecule.
- This leads to better skin absorption and a 1000-fold increase in activity compared with triamcinolone itself.
- If the compounds are injected under the skin, they have equal activity.

• Fluocinolone acetonide, fluocinonide and flunisolide are clinical agents that contain the same acetonide group.



- Good skin absorption can also be achieved by **esterifying** one or more alcohol groups. The corresponding phosphate esters were less active, providing further evidence that lipophilicity is important to the activity of topically applied anti inflammatories.
- Glaxo used this strategy to develop the clinically useful agents betamethasone 17-valerate, betamethasone dipropionate and beclometasone dipropionate.

Betamethasone 17-valerate

Betamethasone 17,21-dipropionate

Beclometasone 17,21-dipropionate

□ 21-Deoxysteroids

- Removal of the 21-OH group from cortisol eliminates activity, but introducing an **extra double bond** in the A ring along with substituents at C6 and C9 restore the activity and results in **fluorometholone**.
- Esterification of the 17-OH group results in better skin absorption and increased topical activity, e.g. **21-dexoybetamethasone 17-dipropionate**.
- **Introducing a halogen** at position 21 was particularly beneficial for the 17-esters. The best activity was obtained using F or Cl, with short chain esters at C-17.
- The best compound arising from these studies was <u>clobetasol</u> propionate.

21-Deoxybetamethasone 17-propionate

Clobetasol 17-propionate

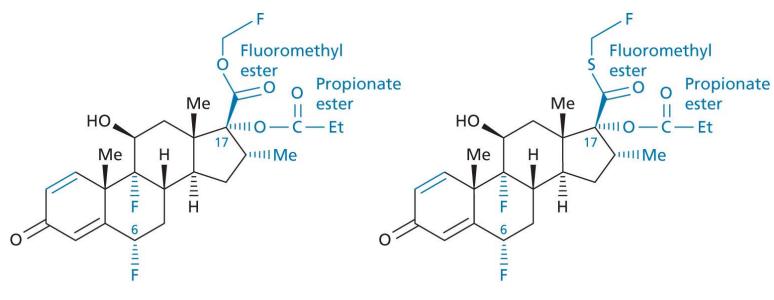
☐ 11-Ketosteroids

- In general, replacing the 11β -OH group of cortisol with a keto group results in a drop in activity.
- However, activity can be **restored** by introducing suitable substituents elsewhere.
- Halogens at positions C-9 and C-21 are particularly important in this respect, e.g. clobetasone butyrate.

☐ Analogues with modified C-17 side chains

- The two carbon chain at C-17 is generally important for activity, but it was found that activity could be retained if the side chain was replaced with a **carboxylic acid** as long as both it and the 17-OH group **were esterified**.
- If only one was esterified then there was no activity.
- This discovery is important that the di-esters would be active at the site of the administration but would be hydrolyzed to inactive compounds as soon as they reached the blood circulation.
- Thus reducing the chances of unwanted side effects elsewhere in the body.

- A variety of esters were synthesized which demonstrated that the 17α -propionate and 17β -fluoromethyl esters were ideal.
- Further variations led to the discovery that the 17β -fluoromethyl thioester was also beneficial leading to the clinically important fluticasone propionate.
- This agent has a high affinity for target receptors, high potency, and low oral bioavailability (1%) because of low solubility and rapid metabolism in the liver.



☐ Glucocorticoids used in asthma treatment

- Glucocorticoids are used as anti-inflammatory agents in the treatment of asthma and are administered by inhalation in order to reduce the risks of side effects caused by their presence in the blood supply.
- Most of the glucocorticoids used in asthma treatment are rapidly metabolized in the liver.
- Of more significance is the proportion of inhaled dose that gets absorbed into the blood supply through the lungs.
- Therefore, it is important that glucocorticoids used in asthma treatment are susceptible to metabolic deactivation in the blood; e.g. by esterase.

• <u>Beclometasone dipropionate</u> represented a breakthrough in asthma treatment and is currently used as an inhaler, as are <u>budesonide</u>, <u>ciclesonide</u>, <u>mometasone furoate</u>, and <u>fluticasone propionate</u>.

- Budesonide is an example of a new generation of non-halogenated glucocorticoids. May be expected a drop in activity as a result of a lack of halogen but the nature of **acetal** is key in providing high topical anti-inflammatory activity.
- The acetal group increases the hydrophobic nature of the compound leading to prolonged residence in lung tissue.
- Budesonide has been found to have high receptor affinity and a higher anti-inflammatory potency than fluticasone propionate.
- In contrast, its systematic glucocorticoid activity is 4-7 times lower owing to extensive first-pass metabolism in the liver by the cytochrome P450 enzyme (CYP3A4) to much less potent metabolite.

- <u>Ciclesonide</u> is the latest in this series and is an example of a soft steroid.
- The structure acts as prodrug and is activated by esterases in lung tissue which hydrolyze the C-21 ester to reveal a free alcohol group.
- This is the active compound and has a prolonged duration of action in lung tissue, and a negligible activity elsewhere in the body, despite it being able to reach the circulatory system. This is because it is rapidly metabolized by CYP450 enzymes to inactive metabolites.
- The use of heterocyclic esters at C-17 in <u>mometasone furoate</u> results in high topical anti-inflammatory activity.

☐ Glucocorticoids used in ophthalmology

• A number of steroids have been used as topical anti-inflammatory agents in ophthalmology such as <u>dexamethasone</u>, <u>fluorometholone</u>, <u>betamethasone</u> <u>sodium</u> <u>phosphate</u>, <u>hydrocortisone</u> <u>acetate</u>, <u>prednisolone acetate</u>, & <u>prednisolone sodium phosphate</u>.