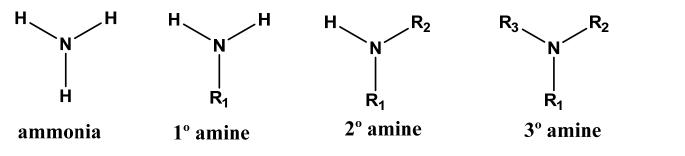


Topic 6: Drug functional groups

Dr. Mohyeddin Assali
PhD. Medicinal chemistry-Nanomedicine
Department of pharmacy
Faculty of Medicine & Health sciences

1. Nitrogen Containing Functional Groups

- The amine group is one of the most important groups responsible for pharmacological activity.
- When it bears a positive charge it can exert biological effects through electrostatic interactions with negatively charged groups within proteins.
- Amines are derived from ammonia and have the structures of mono-, di- and tri- substituted nitrogen.
- The simplest R is CH₃ but usually in drug molecules it is a much larger organic structure.



119

- Amines can carry a positive charge when they accept a proton depending on the pH.
- > At pH values between 0 and 14.

Nitrogen generally present in drug molecules as basic amine and as nitrogens which do not accept a proton (neutral) such as nitrogens present in amides and indole compounds.

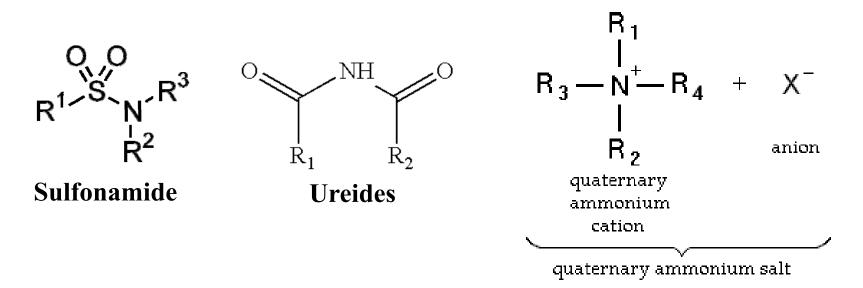
Amide compound

Indole compound

An example of amide is procainamide, the drug structures at pH 0 and pH 14.

- Amide groups contain a carbonyl (or sulfonyl group) next to the nitrogen which pulls electrons away from the nitrogen thus removing its basicity.
- ➤ If more than one carbonyl present adjacent to the nitrogen the nitrogen becomes acidic i.e. it will donate a proton rather than accept one.
- ➤ In the case of sulfonamides and ureides the nitrogen becomes quite strongly acidic.

- Another type of nitrogen is the quaternary nitrogen which is found in some drugs.
- This nitrogen carries a permanent charge and this charge is not affected by the pH.
- ➤ Quaternary ammonium drugs are always in the form of a salt e.g. chloride, bromide, iodide, citrate, mesilate etc.



2. Oxygen Containing Functional Groups

- The behavior of oxygen is more straightforward than nitrogen
- > Oxygen containing compounds can be neutral or acidic but not basic.
- For example carboxylic acids are acidic, phenols are weakly acidic and alcohols are neutral.

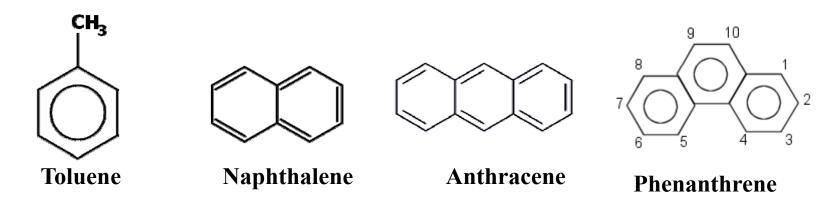
 ph 1

 ph 14

p-hydroxymandelic acid

- > Oxygen containing groups have another two groups which are neutral
- An ether which is neutral and stable to high pH
- •An ester which is neutral but unstable to high pH.
- Ethers are also found in some drugs where sulfur is present instead of oxygen and these are termed thioethers.

3. Aromatic Hydrocarbons



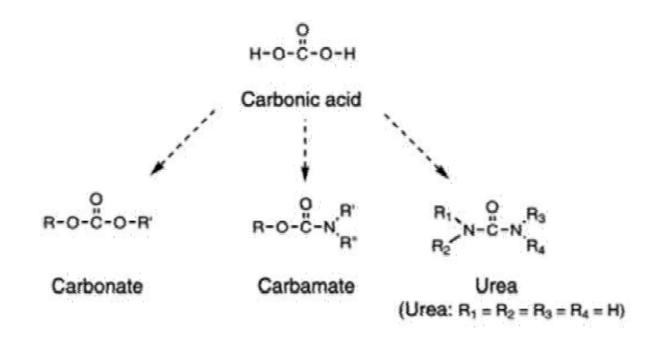
4. Phenols

Structure	Common name	IUPAC name
он	Carbolic acid	Phenol
СН₃	o-Cresol	2-Methylphenoi
O ₂ N-_OH	p-Nitrophenol	4-Nitrophenol
НО	Catechol	1,2-Dihydroxybenzene
—ОН	Resorcinol	1,3-Dihydroxybenzene
но-С	Hydroquinone	1,4-Dihydroxybenzene

5. Functional Derivatives of Carboxylic Acids

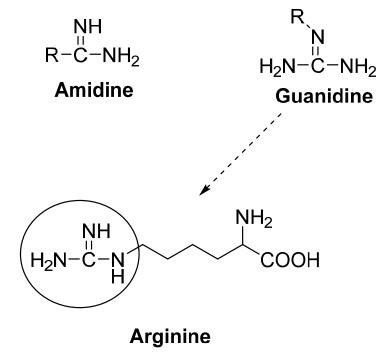
- > Ester and amides
- The cyclization of an amine and a carboxylic acid. The resulting amide is known as a lactam. Lactams make up the nucleus or several classes of important drugs.
- the cyclilization of an oxygen and a carboxylic acid the result is lactone ring which is also important in many drugs.

6. CARBONATES, CARBAMATES, AND UREAS



7. AMIDINES AND GUANIDINES

- Amidines are actually a functional derivative of an amide, while the guanidine is a functional derivative of urea.
- They are found as an integral part of several drug molecules.
- The guanidine moiety is an important unit in the naturally occurring amino acid arginine.



128

8. Sulfonic Acids and Sulfonamides

Benzene sulfonic acid

Methane sulfonic acid

p -Toluene sulfonic acid

Common: Benzene sulfonamide IUPAC: Benzene sulfonamide Common: p -Toluene sulfonamide IUPAC: 4-Methylbenzene sulfonamide

Common: p-Aminobenzene sulfonamide

Sulfanilamide

IUPAC: 4-Aminobenzene sulfonamide

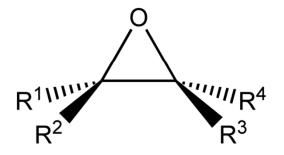
9. Thioethers and Nitro Groups

10. Heterocycles

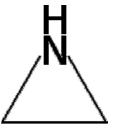
- Heterocycles are defined as cyclic molecules that contain one or more heteroatoms in a ring.
- A heteroatom is an atom other than carbon. It would be impossible to introduce all of the possible heterocycles that are of medicinal value so a selected number of monocyclic, bicyclic, and tricyclic rings

10.1. THREE-MEMBERED RING HETEROCYCLES

A saturated three-membered ring containing oxygen is known as the oxirane, the presence of these molecules also referred as epoxide.



➤ A saturated three-membered ring containing nitrogen is known as the azridine ring.



10.2. FOUR-MEMBERED RING HETEROCYCLES

 \triangleright Only one derivative of a four-membered ring heterocycle that will be mentioned here; the β Lactam.

10.3. FIVE-MEMBERED RING HETEROCYCLES

> Two common five-membered ring heterocycles are furan and

tetrahydrofuran.

Common:

IUPAC:

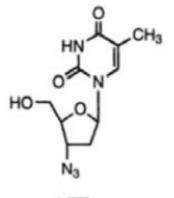
Furan

Oxole

Tetrahydrofuran (THF)

Oxolane

Furosemide



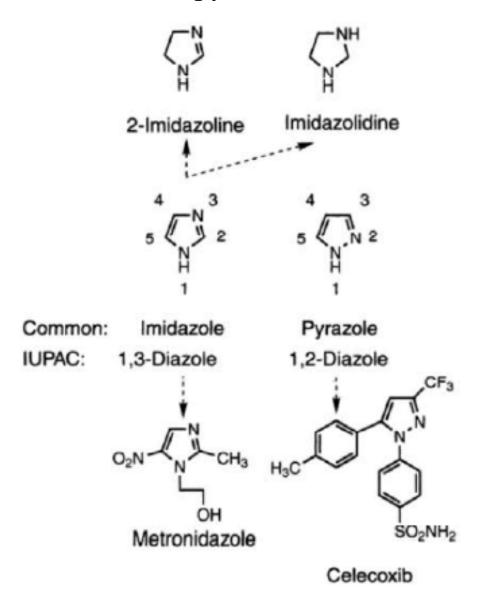
AZT

Azidothymidine (zidovudine)

Two common five-membered ring hetrerocycles containing nitrogen are pyrrole and pyrrolidine

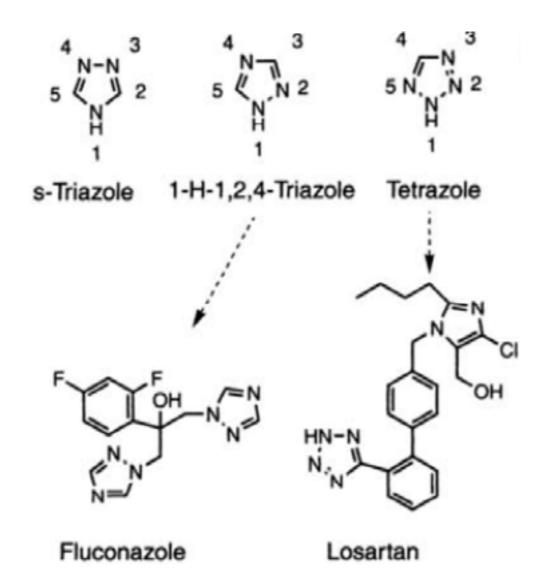
Two sulfur-containing heterocycles, thiophene and tetrahydrothiophene

Two important dinitrogen heterocycles are found in medicinal agents; these are imidazole and pyrazole

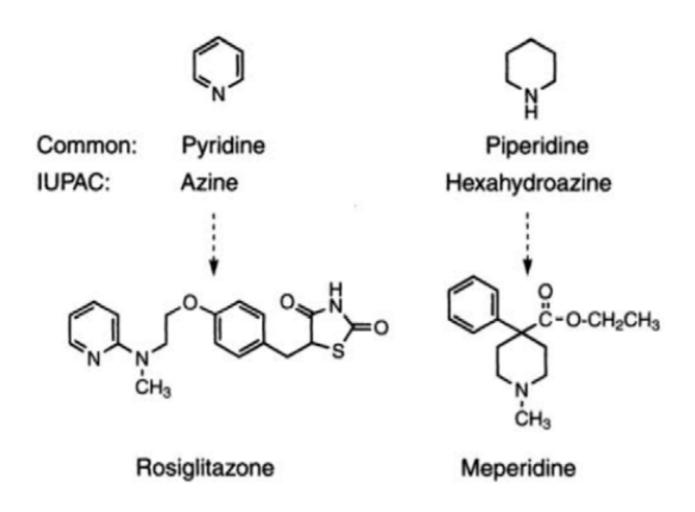


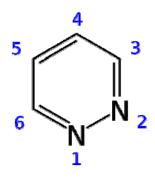
➤ Heterocycles with nitrogen and sulfur called 1,3-Thiazole

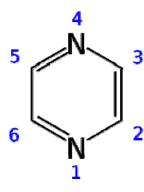
➤ Complex Five-Membered Heterocycles



10.4. SIX-MEMBERED RING HETEROCYCLES



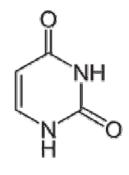




pyridazine

pyrimidine

pyrazine



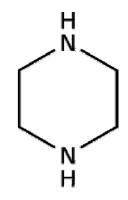


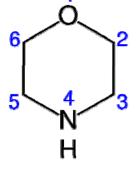
Uracil

Thymine

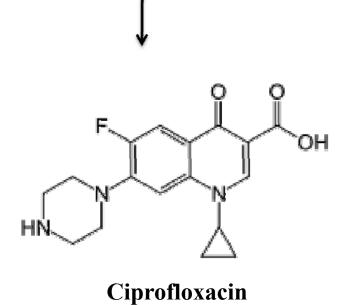
Cytosine

10.5. SATURATED SIX-MEMBERED HETEROCYCLES





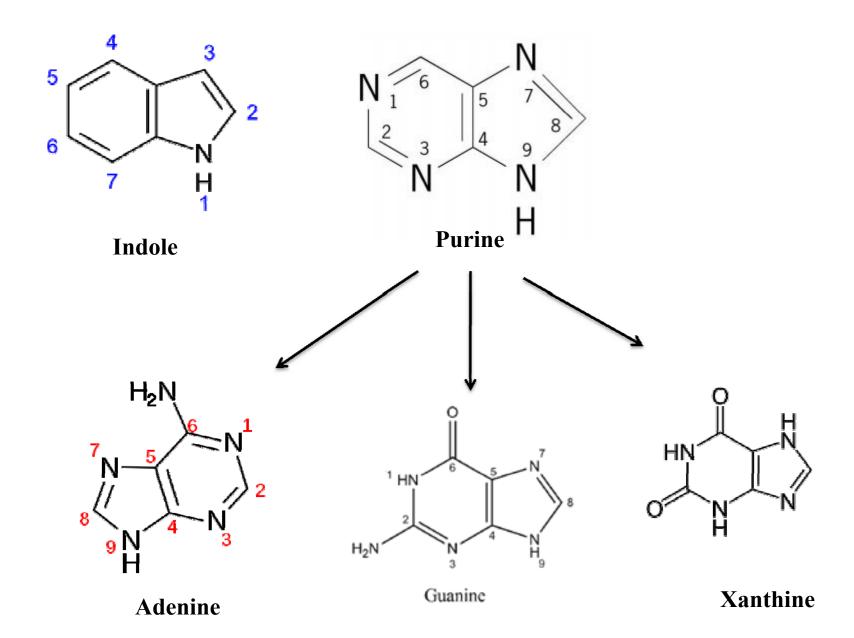
Piperazine

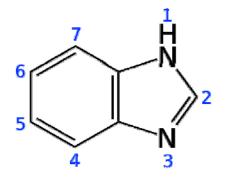


Morpholine

Phenmetrazine

10.6. BICYCLIC HETEROCYCLES RINGS

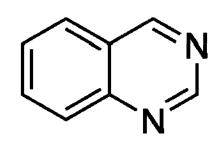




Benzimidazole



Benzothiazole



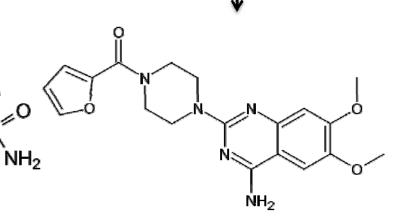
Quinazoline



Omeprazole







Prazocin

10.7. TRICYCLIC HETEROCYCLES RINGS

Phenothiazine